



**Development of CRMs for
Melamine Determination in Milk
Powder and Food**

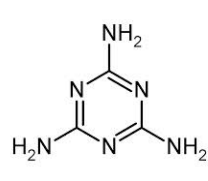
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Introduction

- Melamine (1,3,5-triazine-2,4,6-triamine, $C_3H_6N_6$) was first synthesized in 1834 and had later been found to have extensive uses in the industry
- With high nitrogen content in the melamine molecule, the chemical could be unethically added to food products in order to increase the apparent protein content
- Melamine incidents that were due to tainted pet food in the United States in 2007 and tainted milk powder in China in 2008 have caused significant impact to the analytical chemistry world
- After the crises, the analysis of melamine becomes one of the routine measurements for food and laboratory testing. A maximum residue limit (MRL) of 1 mg/kg was set up for melamine in milk and 2.5 mg/kg for food

Nc1nc(N)c(N)n1

Chemical Formula: $C_3H_6N_6$
Average Mass: 126.12 g/mol
CAS number: 108-78-1
IUPAC name: 1,3,5-triazine-2,4,6-triamine

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Introduction (Cont'd)

Toxicity of melamine

Melamine is not metabolized in human body and is rapidly eliminated in the urine. Thus this compound has a low acute toxicity. However, long term of high dose exposure can cause the formation of bladder stones and crystals in the urine.

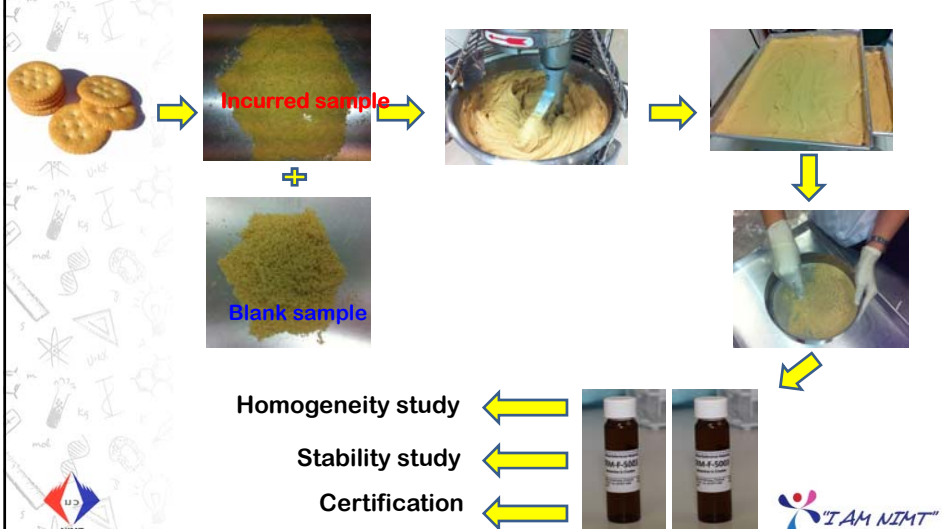
Thai Food and Drug Administration (Thai FDA) has set a regulation limits of melamine contamination in milk and related food as

- less than 1mg/kg for all kinds of milk powder
- less than 2.5 mg/kg for dairy products and related food



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Preparation of Candidate CRM: Melamine in Cracker



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Preparation of Candidate CRM: Melamine in Milk Powder

Blank sample **methanol**

Homogeneity study ←

Stability study ←

Certification ←

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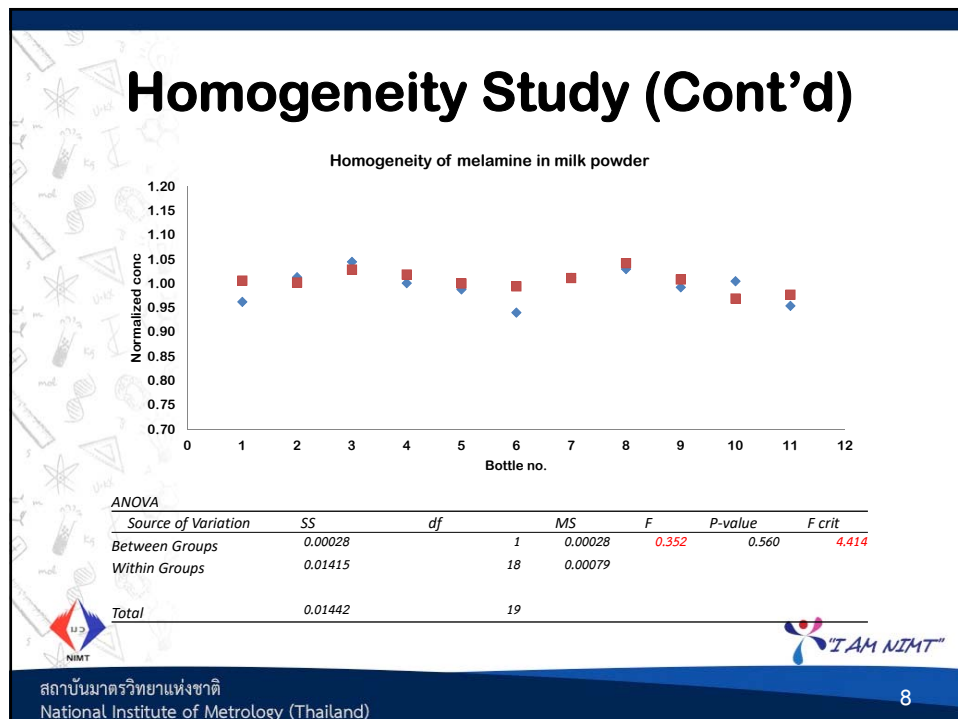
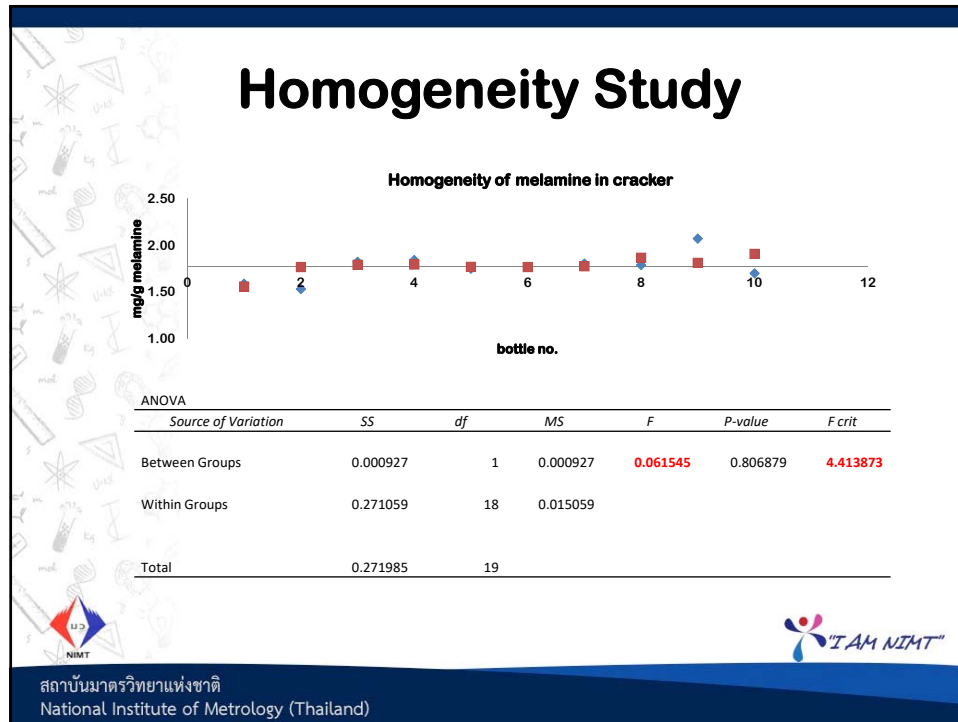
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Homogeneity study

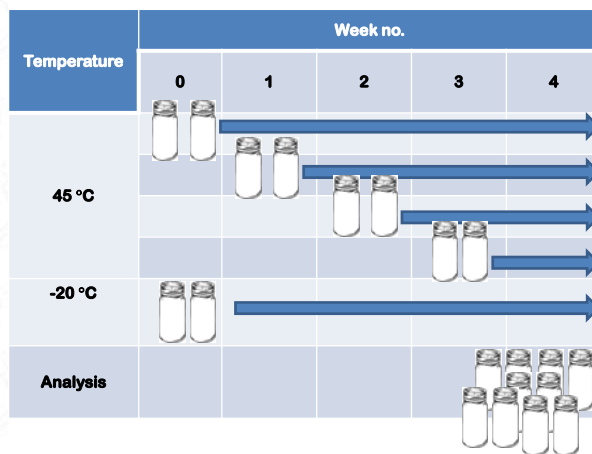
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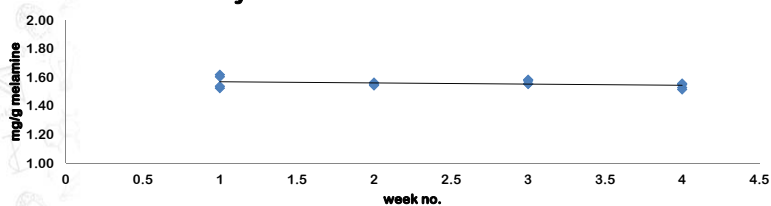
Short-term stability-isochronous scheme



Analyze all samples under repeatability conditions

Short-term stability study

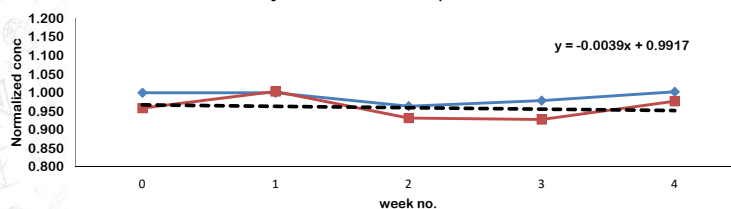
Stability of melamine in cracker at 45 oC



Statistical parameter	Value
Slope (b), $\mu\text{g/g}$	-14.7515
Standard error of slope (s_b), $\mu\text{g/g/week}$	10.42588
Degree of freedom	14
$t_{\text{calc}} = \frac{ b }{s_b}$	1.4149
$t_{\text{crit}} = t(0.05, 14)$	2.145
Statistical significance at 95% CI	No

Short-term stability study (Cont'd)

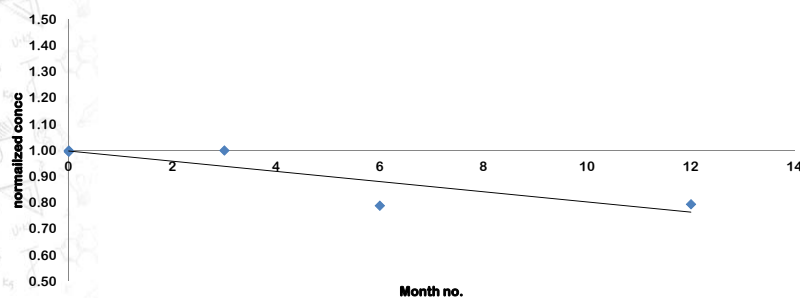
Short term stability of melamine in milk powder stored at 40 °C



Statistical parameter	Value
Slope (b), $\mu\text{g/g}$	-0.0039
Standard error of slope (s_b), $\mu\text{g/g/week}$	0.0067
Degree of freedom	8
$t_{\text{calc}} =$	0.58
$t_{\text{crit}} = t(0.05, 8)$	2.306
Statistical significance at 95% CI	No

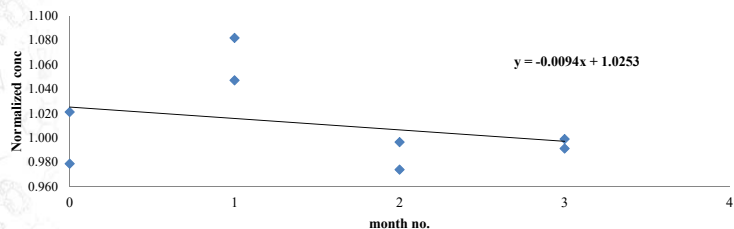
Long-term stability study

Long-term stability of melamine in cracker stored at -20 °C



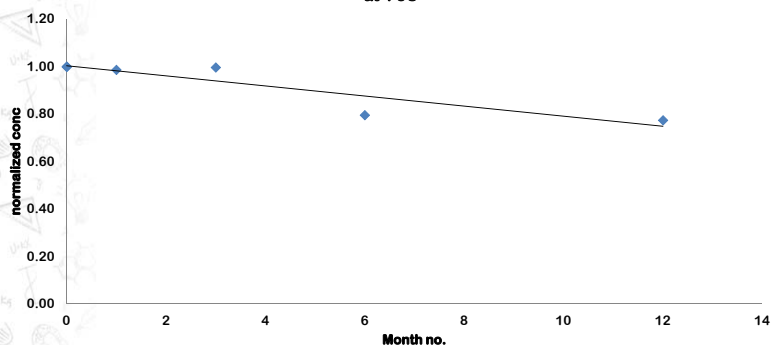
Long-term stability study (Cont'd)

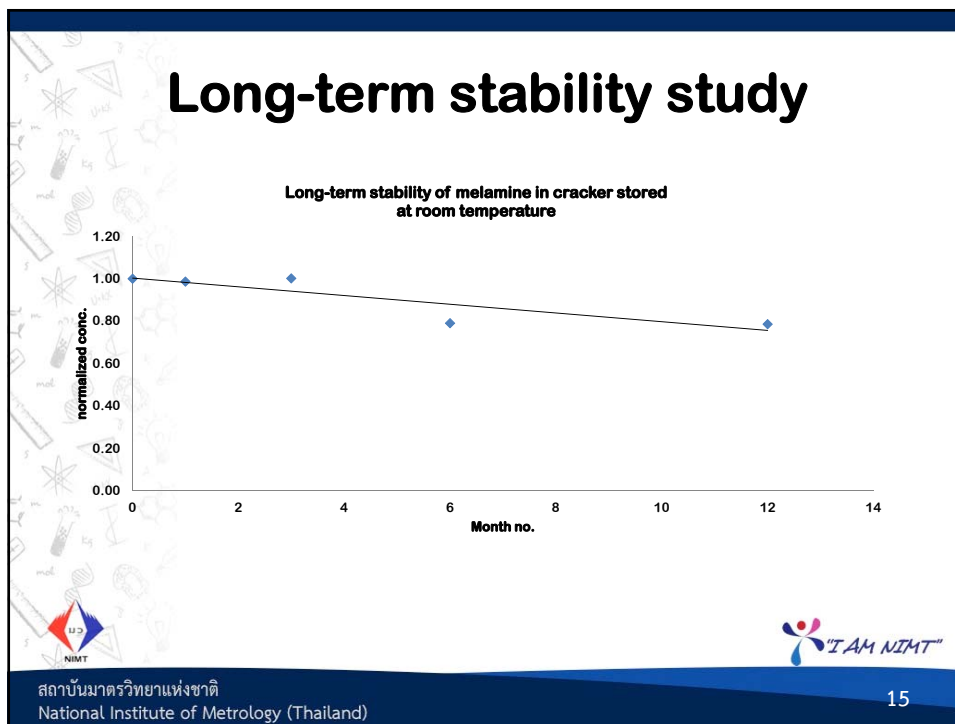
Long term stability of melamine in milk powder stored at room temperature



Long-term stability study

Long-term stability of melamine in cracker stored at 4 oC

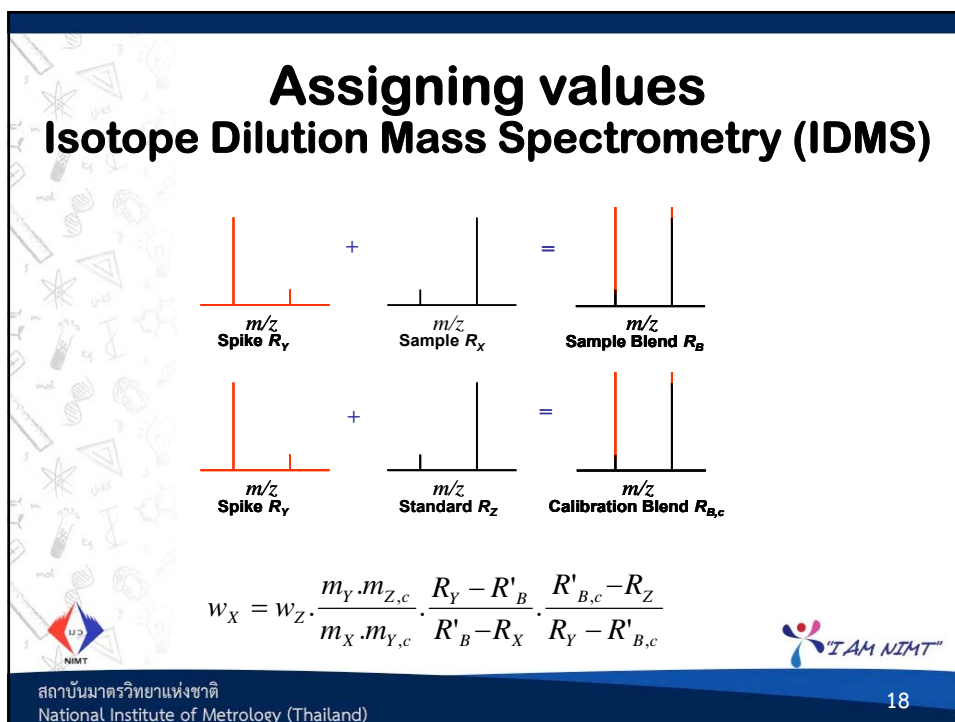
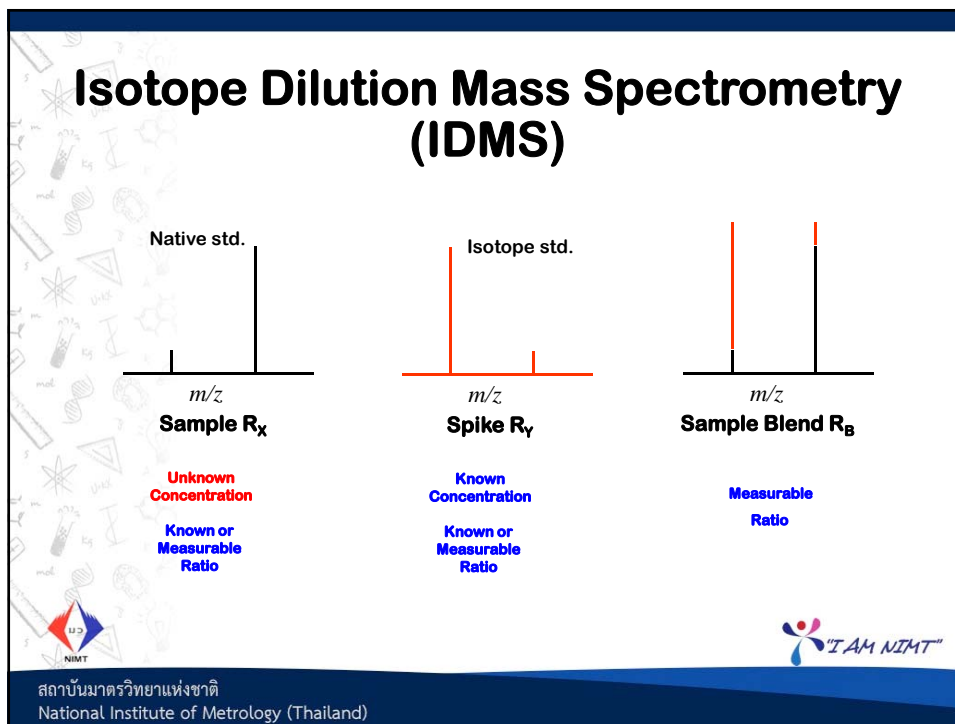




Long-term stability study

Melamine Storage temp	Student's t (tcrit)	Parameters from regression analysis		tcalc	Interpretation
		Slope, abs (b1)	Error of slope s(b1)		
-20 °C	2.3646	0.0379	0.1288	0.294	stability was observed
4 °C	2.2622	0.0426	0.0999	0.426	stability was observed
Room Temp	2.2622	0.0404	0.1066	0.379	stability was observed

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Assigning values

- Modified AOAC method, 990.33, 49.2.17 (1996) with IDMS technique were applied to characterize property values of the CRM

$$w_X = \frac{w_Z \cdot m_Y \cdot m_{Z,c}}{m_X \cdot m_{Y,c}} \cdot \frac{R'_B}{R'_{B,c}}$$

- w_X, w_Z = mass fraction of analyte in the sample and the calibration standard
- m_Y = mass of labelled solution added in the sample blend
- $m_{Y,c}$ = mass of labelled standard solution added in the calibration blend
- m_X = mass of sample in the sample blend
- $m_{Z,c}$ = mass of standard added in the calibration blend
- $R'_B, R'_{B,c}$ = peak area ratio in the sample and calibration blend

Measurement uncertainty

Uncertainty budget for melamine in cracker

Factor	Value x	Uncertainty u(x)	Relative Uncertainty u(x)/x
Measurement equation factor			
Method Precision	1	0.02321	2.32%
$m_{z,c}$	0.14526	0.000071	0.05%
m_Y	0.15513	0.000071	0.05%
$m_{Y,c}$	0.15457	0.000071	0.05%
m_X	1.00513	0.000071	0.01%
w_Z	10.1968	0.110669	1.09%
Additional Factor			
Homogeneity	1	0.007	0.70%
Long term stability	1	0.03	3.00%
Dry mass	1	0.0004	0.0004
Extraction effects	1	0.01	1.00%
Interference from two different ion pairs	1	0.0054	0.54%

Measurement uncertainty

Uncertainty Analysis

Results

$W_x =$	1.563	mg/kg
$u(x) =$	0.065	mg/kg
$u(x)/x =$	4.17%	
$V_{\text{eff}}(\text{total}) =$	129.809	
$k =$	1.98	(@ 95% level)
$U(x) =$	0.129	
$\%U(x) =$	8.24%	

Certified value for melamine in cracker = 1.56 ± 0.13 mg/kg



Conclusions

1. Two matrix CRMs were prepared from incurred and spiked samples in accordance with ISO 17034.
2. Certified values are based on exact matching isotope Dilution mass Spectrometry (IDMS) technique.
3. Certified values for melamine in milk powder and melamine in cracker are 1.03 ± 0.16 mg/kg and 1.56 ± 0.13 mg/kg, respectively.
4. These CRMs can be used for method validation or as a quality control sample



APFAN PT-2 Workshop
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